



Reading and Interpreting Dairy Manure Analyses

Introduction

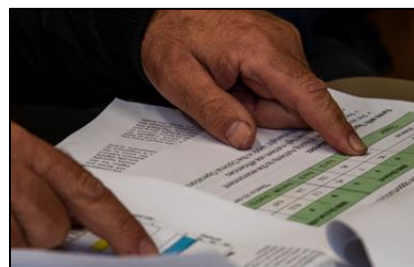
Dairy manure contains all 17 essential nutrients for crop growth and has additional yield enhancing properties. The use of manure as a fertilizer can reduce reliance on synthetic fertilizer, increase soil productivity, enhance yield stability over time, and contribute to increased economic efficiency of dairy farming. However, over-application of manure can have negative impacts on water and air quality.

A manure analysis gives a detailed understanding of the characteristics and value of this resource. This can allow manure use to be optimized, and undesirable environmental impacts minimized (Figure 1). [Agronomy Fact Sheet #38](#) outlines guidance for manure sample collection and submission. In the current factsheet we list what is commonly included in a manure analysis report and explain how to interpret that information.

Manure Analysis

Manure analysis reports should, at a minimum, include organic nitrogen (N), ammonium or inorganic N ($\text{NH}_4\text{-N}$), total N (organic + inorganic N, sometimes referred to as total Kjeldahl nitrogen or TKN), total phosphorus (P), and total potassium (K). This distinction in N forms is important because organic and inorganic N behave differently in soil and contribute differently to crop growth (additional

Fig. 1: Analyzing different manure sources on a farm is essential for determining cropland manure application rates.



information on manure nutrient crediting can be found in [Agronomy Fact Sheet #4](#)).

Most laboratories offer additional testing. See the [Recommended Methods of Manure Analysis](#) for standardized methods for sampling, processing, and analysis of manure samples.

Interpreting Reports

Dairy manure can be categorized as a solid, slurry, or liquid based on the proportion of solids versus liquid in the manure. Manure analysis reports will report either percent solids or percent moisture. A manure sample with 12% solids has 88% moisture, and vice versa. The terms "as-received", "wet basis", and "as is" all refer to manure as it was received by the laboratory.

Nutrient content of manure can be reported as a percentage (%), pounds per ton (lbs/ton), pounds per 1000 gallon (lbs/kgal), or parts per million (ppm). Pounds per 1000 gallons is used for liquid and slurry manure (less than 10%

Table 1: Common conversions of units reported in manure analysis (adapted from Recommended Methods of Manure Analysis).

Conversion	Converting back
Percent (%) $\times 83.4 = \text{lbs}/1000 \text{ gal}$	$\text{lbs}/1000 \text{ gal} \div 83.4 = \text{Percent} (\%)$
Percent (%) $\times 20 = \text{lbs}/\text{ton}$	$\text{lbs}/\text{ton} \div 20.00 = \text{Percent} (\%)$
Percent (%) $\times 10,000 = \text{ppm}$	$\text{ppm} \div 10,000 = \text{Percent} (\%)$
$\text{ppm} \times 0.00834 = \text{lbs}/1000 \text{ gal}$	$\text{lbs}/1000 \text{ gal} \div 0.00834 = \text{ppm}$
$\text{ppm} \times 0.002 = \text{lbs}/1000 \text{ gal}^*$	$\text{lbs}/1000 \text{ gal} \div 0.002 = \text{ppm}$
$\text{US ton} \times 2000 = \text{lbs}$	$\text{lbs} \div 2000 = \text{US ton}$
$\text{Nitrate N} \times 4.43 = \text{Nitrate}$	$\text{Nitrate} \div 4.43 = \text{Nitrate N}$
$\text{Ammonia N} (\text{NH}_4\text{-N}) \times 1.22 = \text{Ammonia} (\text{NH}_4)$	$\text{Ammonia} (\text{NH}_4) \div 1.22 = \text{Ammonia N} (\text{NH}_4\text{-N})$
$\text{Phosphorus (P)} \times 2.29 = \text{P}_2\text{O}_5$	$\text{P}_2\text{O}_5 \div 2.29 = \text{Phosphorus (P)}$
$\text{Potassium (K)} \times 1.20 = \text{K}_2\text{O}$	$\text{K}_2\text{O} \div 1.20 = \text{Potassium (K)}$

*assuming an average density of 8.34 lbs/gal

Table 2: Typical values for dairy manure in New York.

Manure type	Nutrient	Median	Lower range	Upper range
Liquid (lbs/1000gal) <4% solids	Total N	15	11	19
	NH ₄ N	9	6	11
	Organic N	6	5	8
	P ₂ O ₅	5	2	8
	K ₂ O	18	13	23
Slurry (lbs/1000gal) 4-10% solids	Total N	20	18	24
	NH ₄ N	10	9	12
	Organic N	10	9	12
	P ₂ O ₅	8	6	10
	K ₂ O	22	18	27
Semi-solid (lbs/ton) 11-20% solids	Total N	6	5	7
	NH ₄ N	2	2	3
	Organic N	4	3	4
	P ₂ O ₅	3	2	3
	K ₂ O	5	5	7
Solid (lbs/ton) >20% solids	Total N	7	6	10
	NH ₄ N	2	1	3
	Organic N	6	5	7
	P ₂ O ₅	3	2	6
	K ₂ O	7	6	8

solids). Pounds per ton is used for semi-solid and solid manure (more than 10%). Parts per million is commonly used to report micronutrient content of a manure source. Some laboratories report phosphorus as P and as P₂O₅, and potassium as K and K₂O. Use the P₂O₅ and K₂O values when deriving a manure application rate to meet all or a portion of crop nutrient needs. Table 1 shows useful conversions between different ways that manure analyses are reported. For some conversions, the density of the liquid and slurry manure is required. If this is not given in the report, an average density of 8.34 lbs per gallon for liquid dairy manure can be used.

Typical Ranges

Typical manure nutrient values for 240 New York dairy manure samples submitted to [DairyOne](#) are shown in Table 2. This table shows that untreated liquid manure (4-10% solids) had a median of 8 lbs P₂O₅ per 1000 gallons.

In Summary

Manure analysis is essential to understand the nutrient composition of a manure sources, and for planning appropriate application rates. It is important to understand how results are

reported by a laboratory (as is or on DM basis), and what units and metrics are used (P or P₂O₅, etc.) to utilize results.

Additional Resources

- Recommended Methods of Manure Analyses: <https://conservancy.umn.edu/handle/11299/227650>
- Cornell Agronomy Fact Sheet Series: <http://nmisp.cals.cornell.edu/guidelines/factsheets.html>

Disclaimer

This fact sheet reflects the current (and past) authors' best effort to interpret a complex body of scientific research, and to translate this into practical management options. Following the guidance provided in this fact sheet does not assure compliance with any applicable law, rule, regulation or standard, or the achievement of discharge levels from agricultural land.

For more information



Cornell University
Cooperative Extension

Nutrient Management Spear Program
<http://nmisp.cals.cornell.edu>

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